

# REPORT DOCUMENTATION PAGE

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MEMORANDUM FOR PRS (In-House/Contractor Publication)

FROM: PROI (STINFO)

16 Mar 2001

SUBJECT: Authorization for Release of Technical Information, Control Number: **AFRL-PR-ED-AB-2001-055**  
Vaghjiani, Ghanshyam L. (ERC), "Investigations of Chemiluminescence in the CH<sub>2</sub> + O Gas Phase Reaction"

**37<sup>th</sup> AIAA/ASME/SAE/ASEE Joint Propulsion Conference & Exhibit**  
**(Salt Lake City, UT, 8-11 July 2001) (Deadline: 31 Mar 2001)**

**(Statement A)**

1. This request has been reviewed by the Foreign Disclosure Office for: a.) appropriateness of distribution statement, b.) military/national critical technology, c.) export controls or distribution restrictions, d.) appropriateness for release to a foreign nation, and e.) technical sensitivity and/or economic sensitivity.

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Signature \_\_\_\_\_ Date \_\_\_\_\_

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3. This request has been reviewed by the STINFO for: a.) changes if approved as amended, b) appropriateness of references, if applicable; and c.) format and completion of meeting clearance form if required

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4. This request has been reviewed by PR for: a.) technical accuracy, b.) appropriateness for audience, c.) appropriateness of distribution statement, d.) technical sensitivity and economic sensitivity, e.) military/national critical technology, and f.) data rights and patentability

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APPROVED/APPROVED AS AMENDED/DISAPPROVED

PHILIP A. KESSEL

Date

Technical Advisor

Space and Missile Propulsion Division

## Investigations of Chemiluminescence in the CH<sub>2</sub> + O Gas Phase Reaction

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The interactions of carbonaceous combustion species in rocket plumes with the atmosphere are thought to play an important role in the production of ultraviolet, visible, and infrared radiation signatures at high altitudes. A detailed understanding of the pertinent chemical reactions that produce the electronically excited species, and of the competing quenching reactions that remove the internal energy in radiation-less processes is needed to accurately calculate plume spectral signatures and absolute radiances (in the short wavelength region), and their temporal/spatial evolution in the high atmosphere. To facilitate these efforts, we are currently carrying out laboratory investigations to elucidate the reaction mechanism(s) in the oxidation of CH, CH<sub>2</sub>, C<sub>2</sub>H, and C<sub>2</sub>O with O-atoms and O<sub>2</sub>. Sufficient exothermicity in CH, CH<sub>2</sub>, and C<sub>2</sub>H reactions (except C<sub>2</sub>H + O) is available to produce CO in one or more of the triplet states (a, a', and d). Even more reaction enthalpy is available in C<sub>2</sub>O reaction(s) to produce higher excited states of CO (e, A, I, and D). Other excited species such as CH(A<sup>2</sup> $\Delta$ ) in C<sub>2</sub>H plus O or O<sub>2</sub>, and OH(A<sup>2</sup> $\Sigma^+$ ) in CH + O<sub>2</sub> reactions are also possible. CO-uv chemiluminescence has previously been identified in C<sub>2</sub>H + O<sub>2</sub> reaction and both CO-uv and CO-vuv in the C<sub>2</sub>O + O reaction. However, no information is available on the product branching ratios of the excited CO states responsible for the emission. Estimates of the branching ratio of CH(A<sup>2</sup> $\Delta$ ) formation in the reactions of C<sub>2</sub>H with O and O<sub>2</sub> can be found in the literature. To our knowledge, triplet CO formation in CH and CH<sub>2</sub> reactions has not yet been positively identified. Fast discharge-flow tube and pulsed-laser photolysis methods have been employed in this work to study the reaction kinetics and chemiluminescence in these reactions. The experimental approach and results of these studies will be presented.